

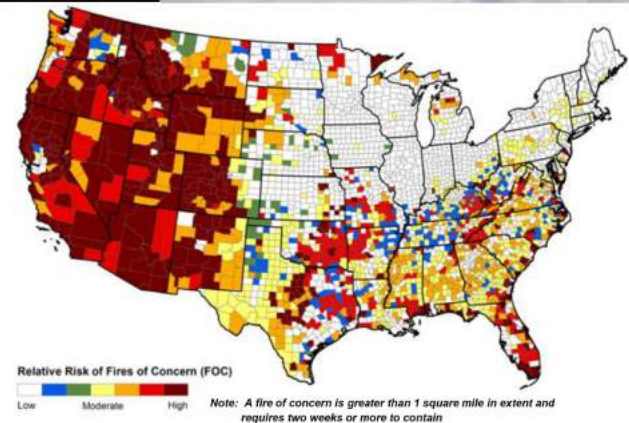
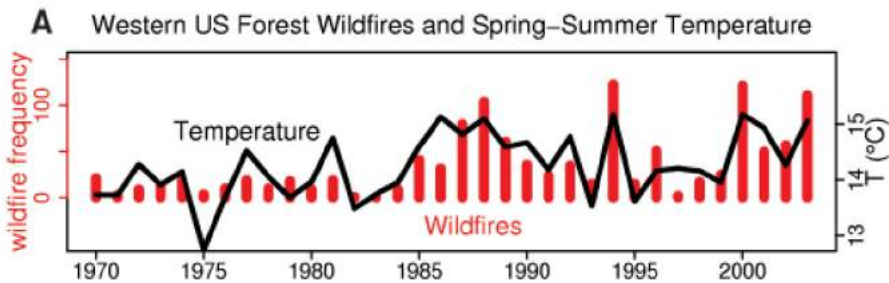
# NOAA FIREX experiment

Fire Influence on Regional and Global Environments Experiment  
(FIREX)

The Impact of Biomass Burning on Climate and Air Quality:  
An Intensive Study of Western North America Fires

## Identification of issue:

A combination of a **warmer, drier climate** with **fire-control practices** over the last century have produced a situation in which we can expect **more frequent fires** and fires of larger magnitude in the Western U.S. and Canada.



**FIREX:** Comprehensive Field Experiment from CIRES/NOAA ESRL Chemical Sciences Division

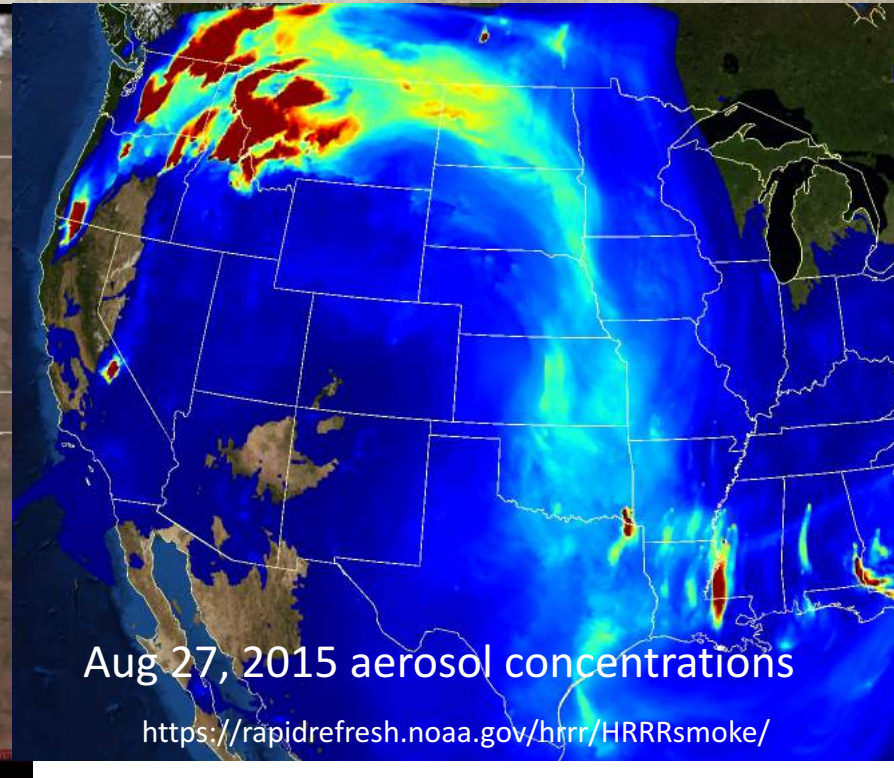
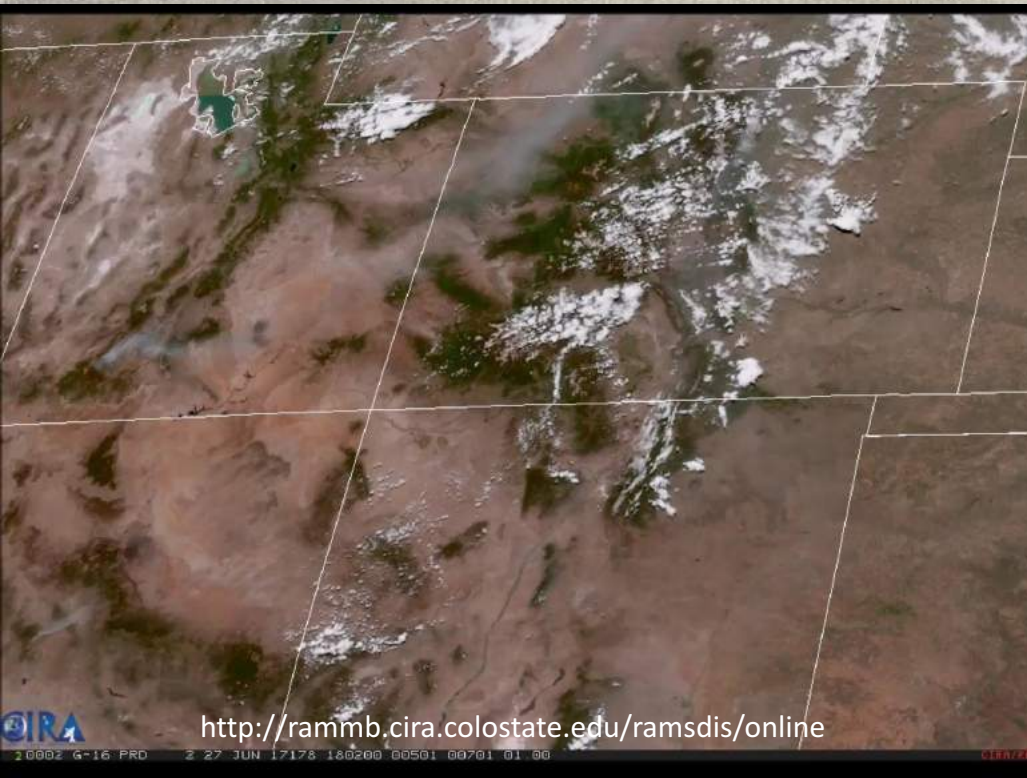
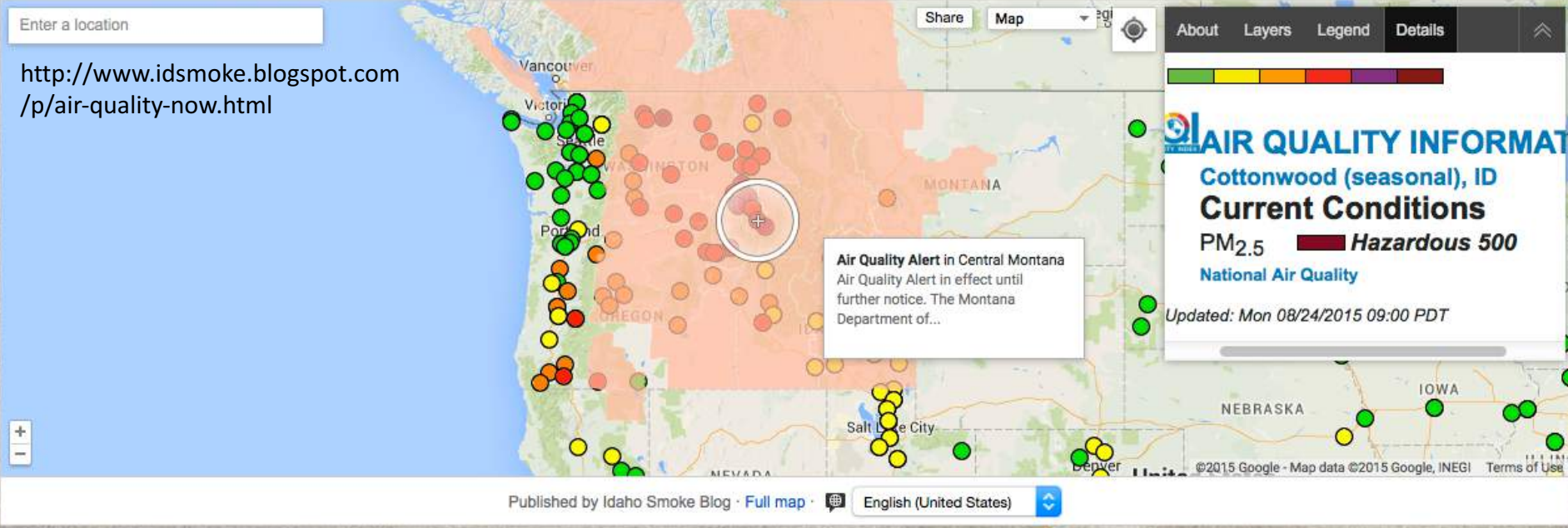
## Contacts:

Carsten Warneke, James Roberts, Joshua Schwarz

NOAA Field and Laboratory Studies during 2015-2019

Enter a location

<http://www.idsmoke.blogspot.com/p/air-quality-now.html>



# Outline

- 1) Planned FIREX activities and timing
- 2) Platforms and sites
- 3) JFSP FASMEE, NASA FIRE-Chem and WE-CAN
- 4) First results of FIREX:

Missoula Fire Lab: Oct 1- Nov

15 2016



NOAA Chemical Sciences Division: Who we are and what do we do



Provide the needed **scientific understanding and information about our atmosphere** to make optimal decisions in the interests of the well being of current and future generations.

Over 100 scientists working on: Air Quality, Climate Change, Ozone Hole  
with  
modeling, laboratory and **small and large scale field work**

Recent field work: Oil&Gas, urban air quality, anthropogenic/biogenic emissions and trends

# Main FIREX Science Questions in 5 Categories

1) What are the **emissions** of gases, aerosols, aerosol precursors, air toxics and greenhouse gases from North American fires?

What is the composition and volatility of the previously unidentified fraction of the emissions?

2) What is the **chemical transformation** of those emissions?

What are the formation mechanisms for secondary species (ozone, SOA and sulfate)?

3) What is **the local air quality** and visibility impact of North American fires?

How important is nighttime smoke for populated areas?

How well do local air quality forecast models work?

4) What are the **regional and long-term impacts** of North American fires on climate?

5) What are the **climate-relevant** properties of BB aerosols?

What role does brown carbon and coatings on black carbon particles play in the optical properties?

What is the composition of PM<sub>2.5</sub>?

# Main FIREX Science Questions in 5 Categories

1) What are the emissions of gases, aerosols, aerosol precursors, air toxics and greenhouse gases from North American fires?

## Detailed Science Questions in White Paper

available at: <http://esrl.noaa.gov/csd/projects/firex/>

Additions from FIRE-Chem, WE-CAN and FASMEE!!!

4) What are the regional and long-term impacts of North American fires?

5) What are the climate-relevant properties of BB aerosols?

What role does brown carbon and coatings on black carbon particles play in the optical properties?

What is the composition of PM<sub>2.5</sub>?

# New science of FIREX

1. **New instrumentation** and satellites
2. **Comprehensive effort** with large science community using ground, mobile, laboratory and aircraft and modeling/forecasting
3. **Simultaneously deployments** of platforms
4. **Nighttime** fires and smoke
5. Years **building knowledge before** large field experiment
6. **Transfer new knowledge** to partners outside US

## Collaborations of FIREX

### Aircraft:

**NASA** DC 8: FIRE-Chem

**NASA** B200: FIRE-Chem

**NCAR/NSF** C130: WE-CAN

### Mobile Labs:

**NOAA** CSD

**Aerodyne**

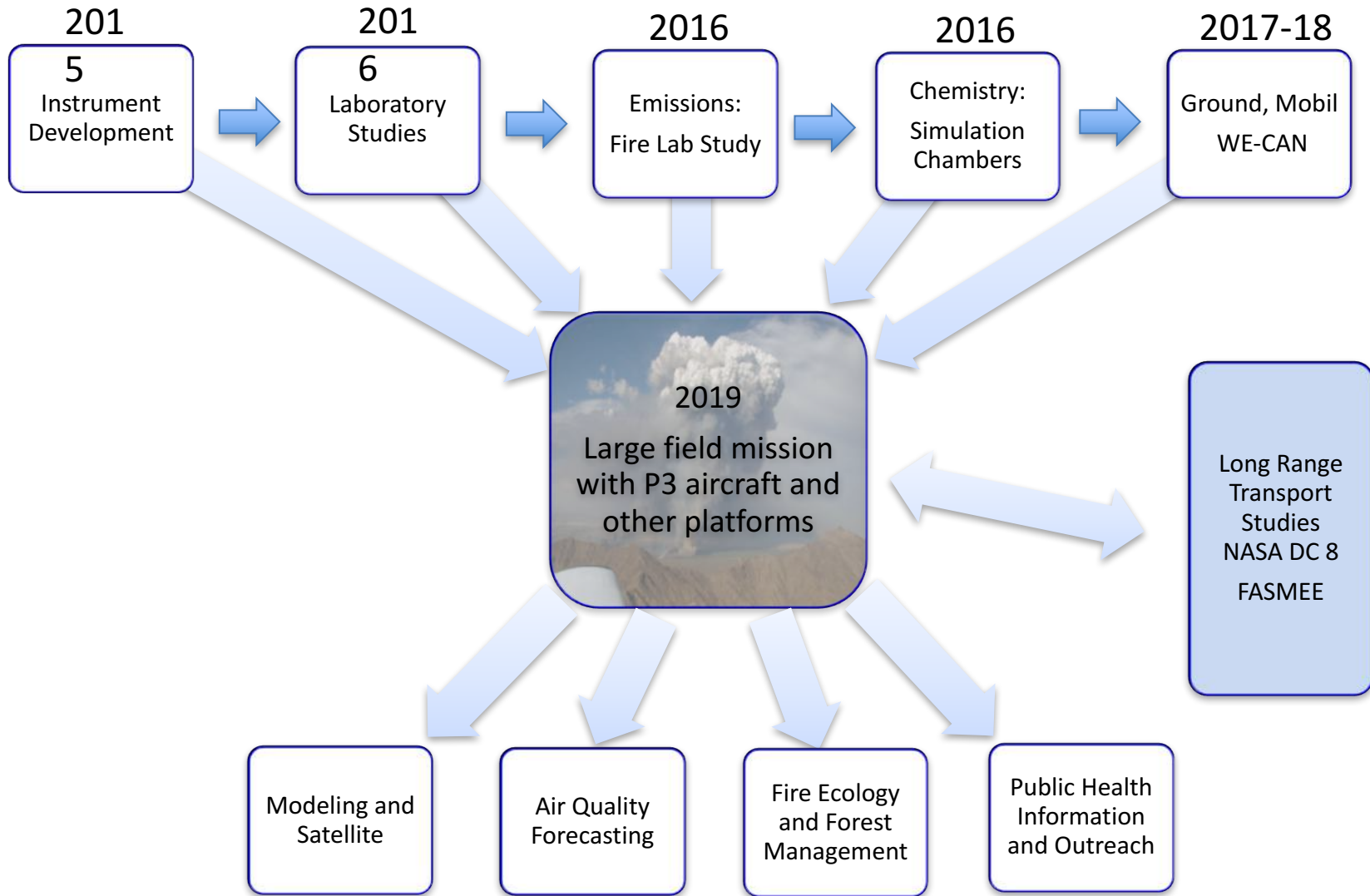
**Joint Fire Science Program: FASMEE project**

several controlled burns in 2018-2021

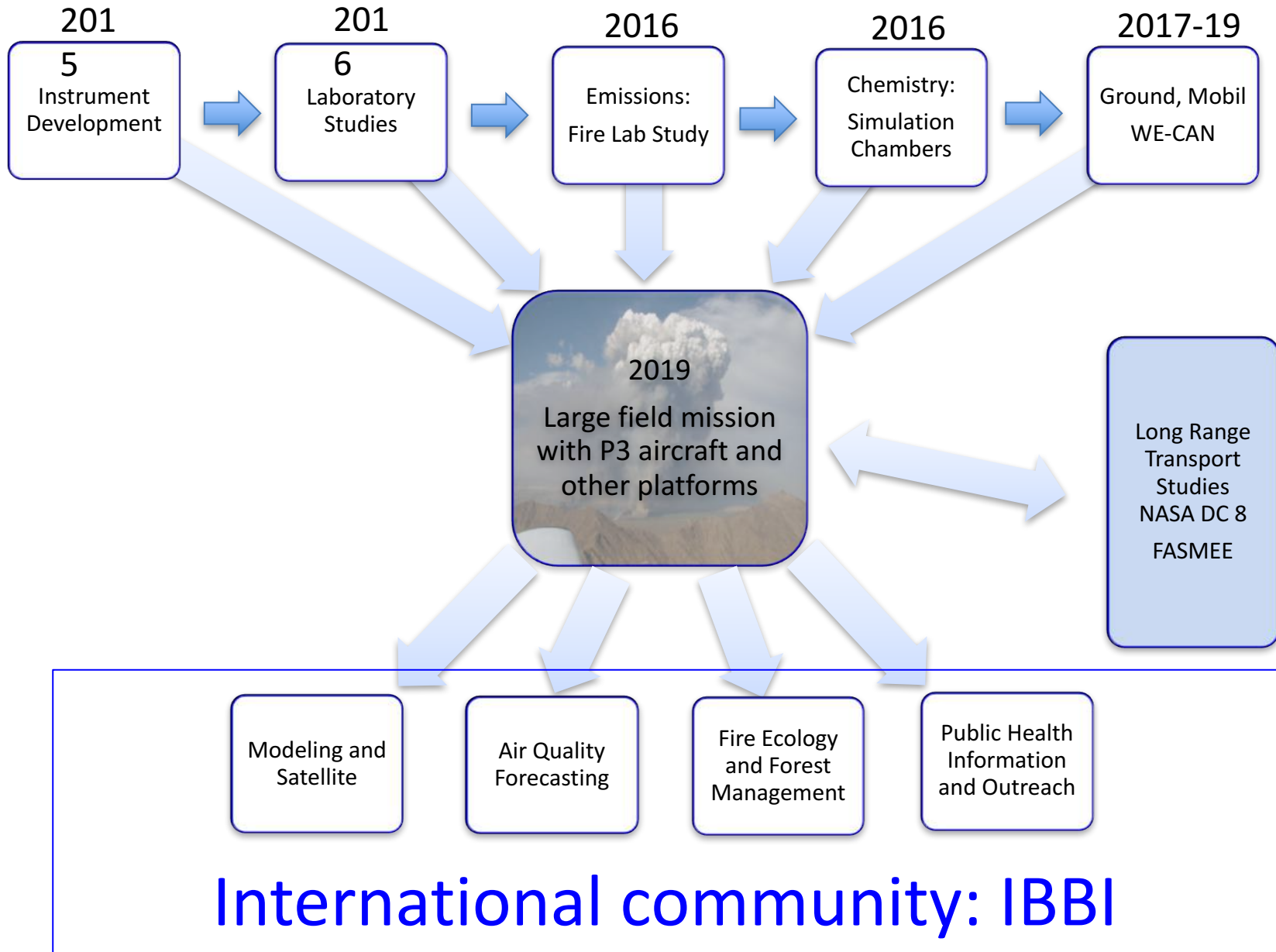
Ground and Lab (TBD/potentially)

**IBBI/IGAC**

# FIREX workflow chart



# FIREX workflow chart





NOAA CSD FIREX Management  
Points of Contact:  
Carsten Warneke, James Roberts, Joshua Schwarz

Carsten Warneke<sup>1,2</sup>

James M. Roberts<sup>1</sup>

Joshua P. Schwarz<sup>1</sup>

Robert J. Yokelson<sup>3</sup>

R. Bradley Pierce<sup>4</sup>

Barry Lefer<sup>5</sup>

James H. Crawford<sup>6</sup>

Kirk R. Baker<sup>7</sup>

Amy P. Sullivan<sup>8</sup>

NOAA WP-3D gas phase measurements

Fire Lab study, field observations with mobile labs

NOAA WP-3D aerosol measurements

Fire Lab study, field observations

Modeling, data assimilation, and satellites

FIREChem coordination

FIREChem coordination

EPA and FASMEE coordination

NSF/NCAR C130 coordination

NOAA ESRL Chemical Sciences Division, Boulder, CO

Cooperative Institute for Research in the Environmental Sciences (CIRES), University of Colorado, and NOAA, Boulder, CO

Department of Chemistry, University of Montana, Missoula, MT

NOAA NESDIS Center for Satellite Applications and Research (STAR), Cooperative Institute for Meteorological Satellite Studies, Madison, WI

NASA Earth Science Division, Tropospheric Composition Program, Washington, DC

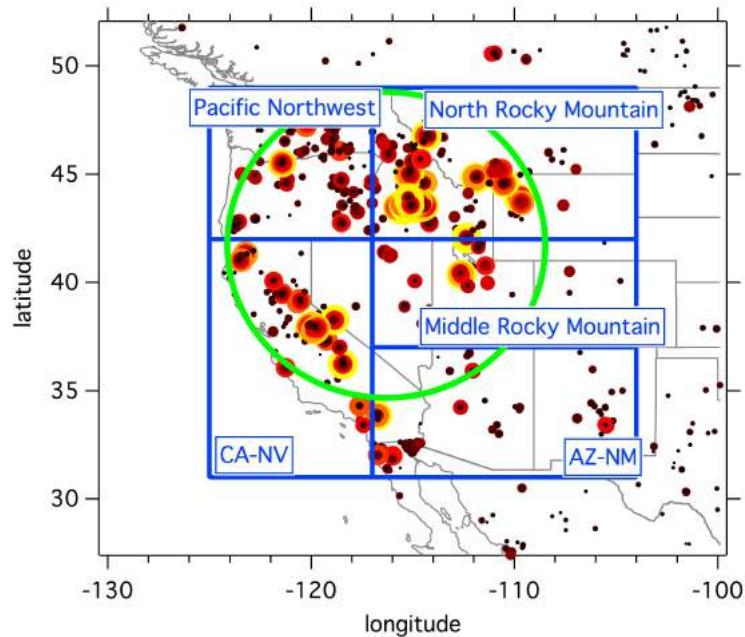
NASA Science Directorate, Chemistry and Dynamics Branch, Langley Research Center, Hampton, VA

U.S. Environmental Protection Agency, Research Triangle Park, NC

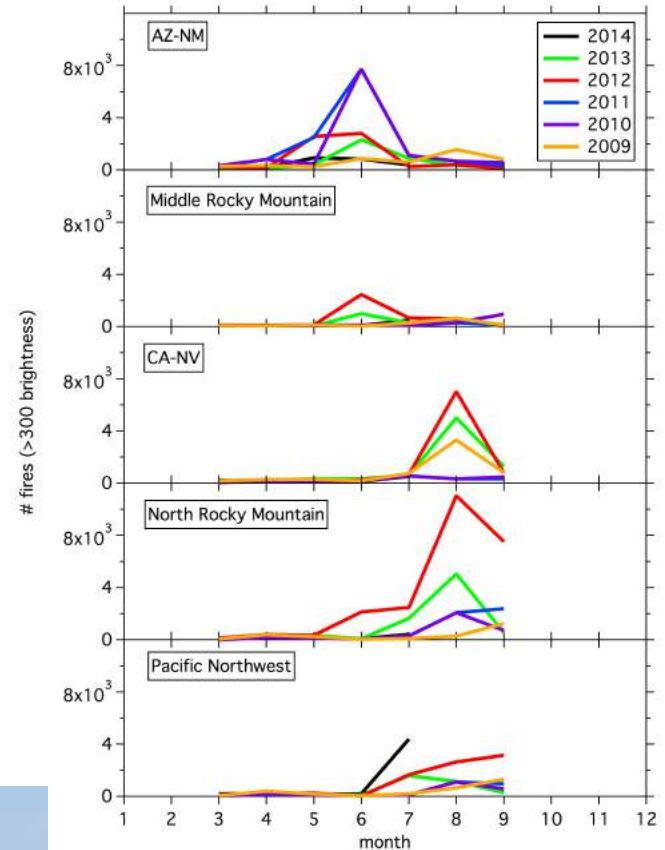
Department of Atmospheric Science, Colorado State University, Fort Collins, CO

NOAA AC4 support of FIREX for several research groups  
and coordination with other funding agencies

# NOAA WP-3 aircraft in fire season 2019



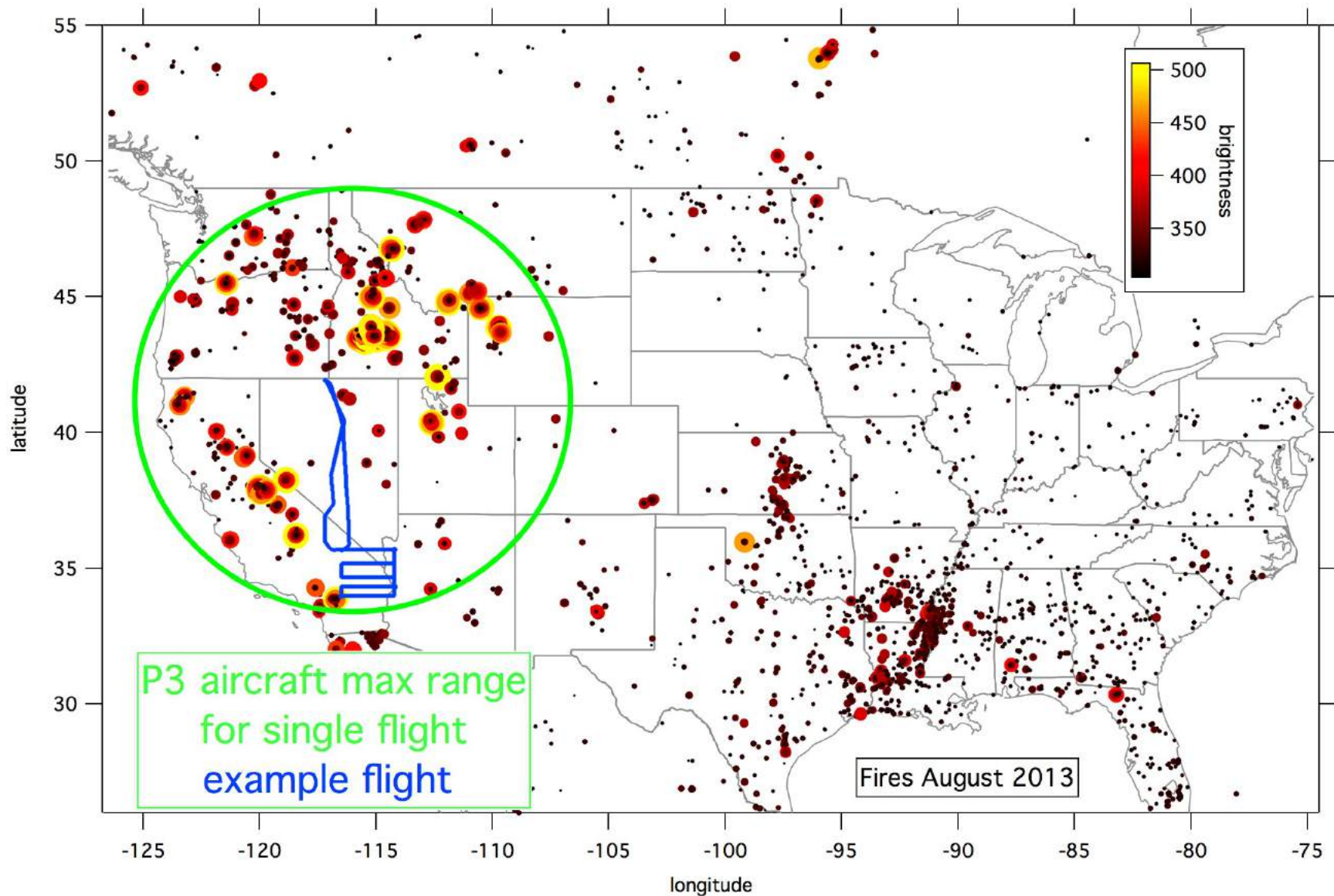
There have always been fires in the NW U.S. in range of the NOAA WP-3 aircraft in summer.



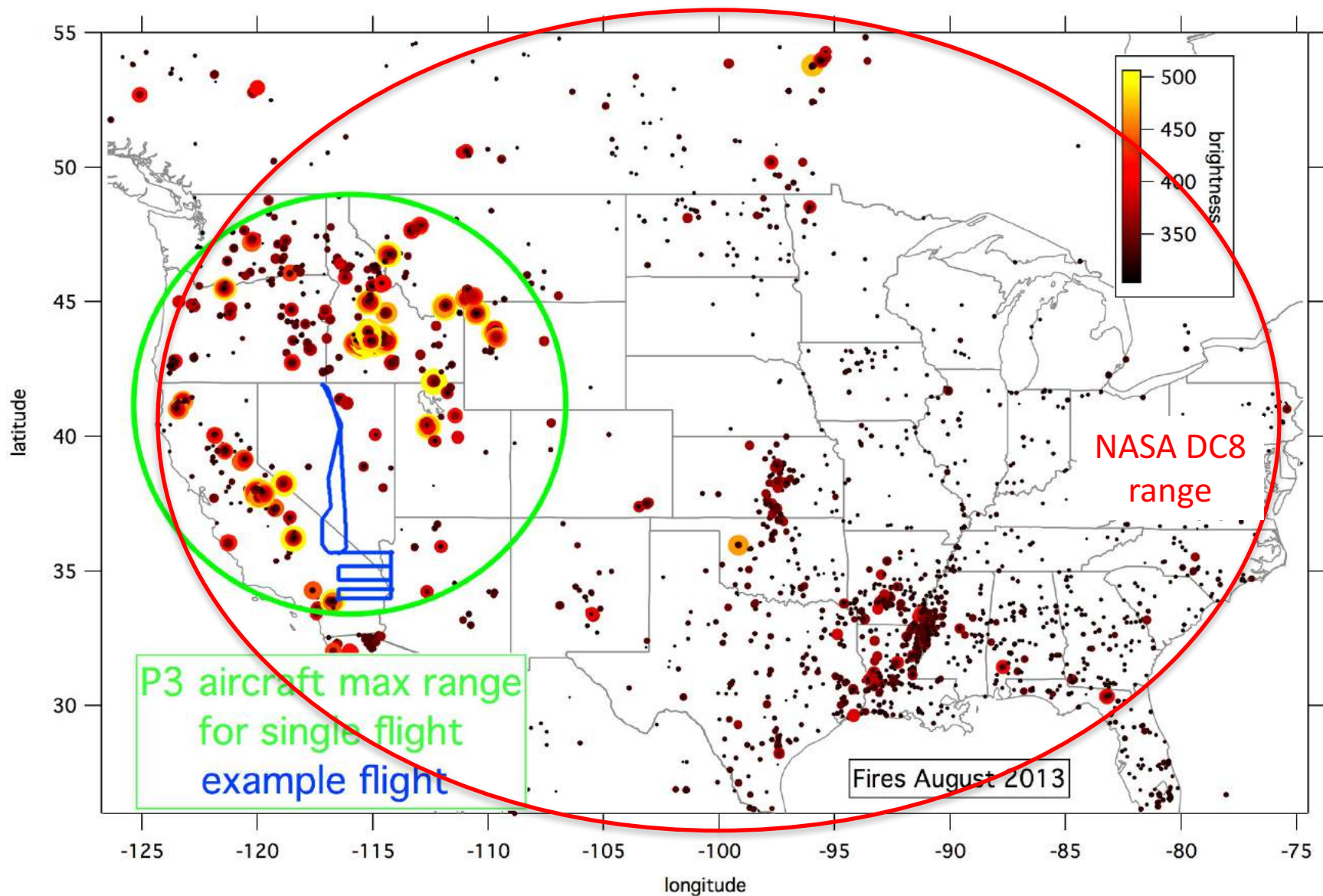
Highly instrumented aircraft used by NOAA CSD since 25 years

often used for hurricane research

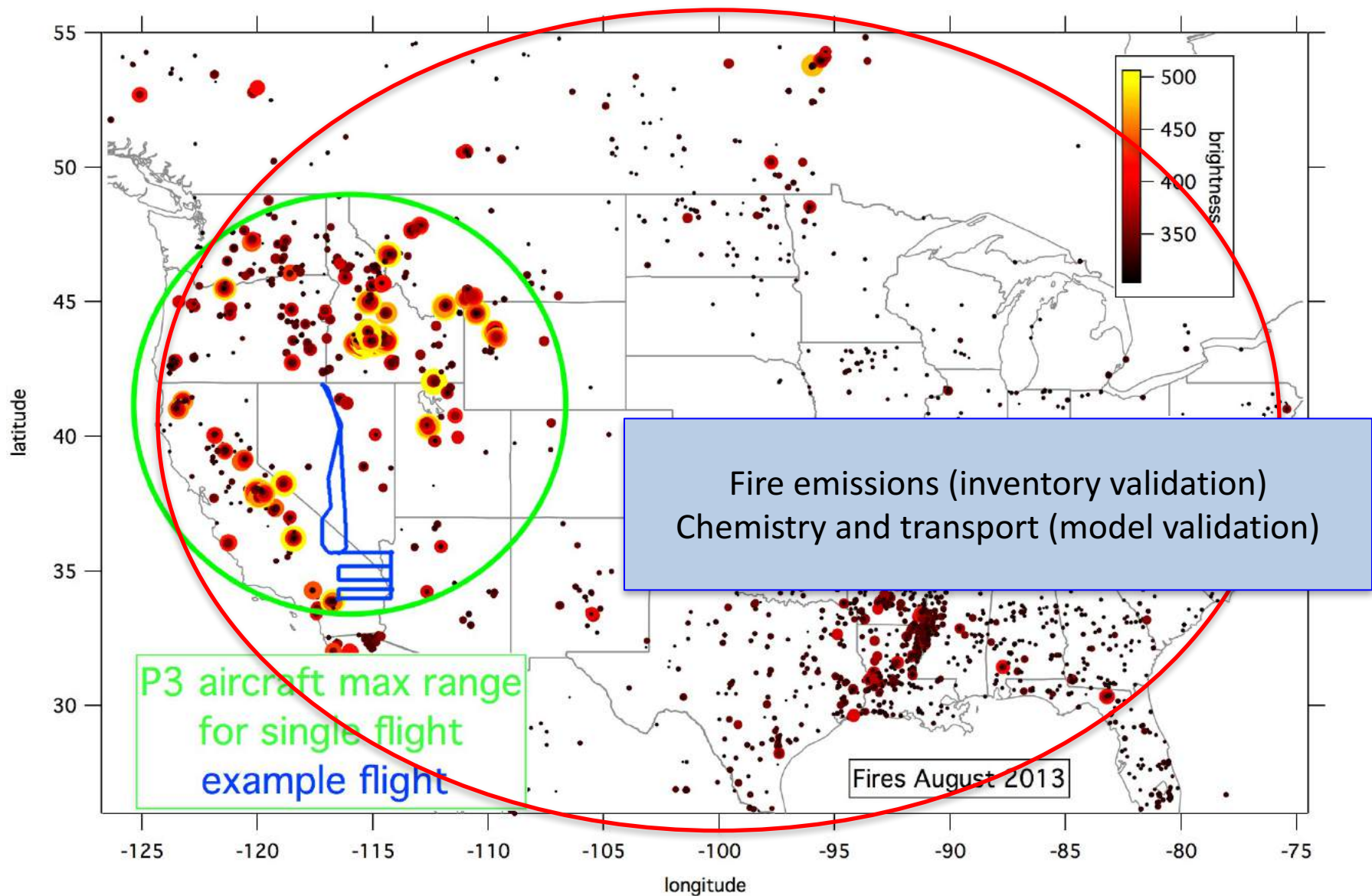
# NOAA WP-3 aircraft in fire season 2019



# NOAA WP-3 aircraft in fire season 2019



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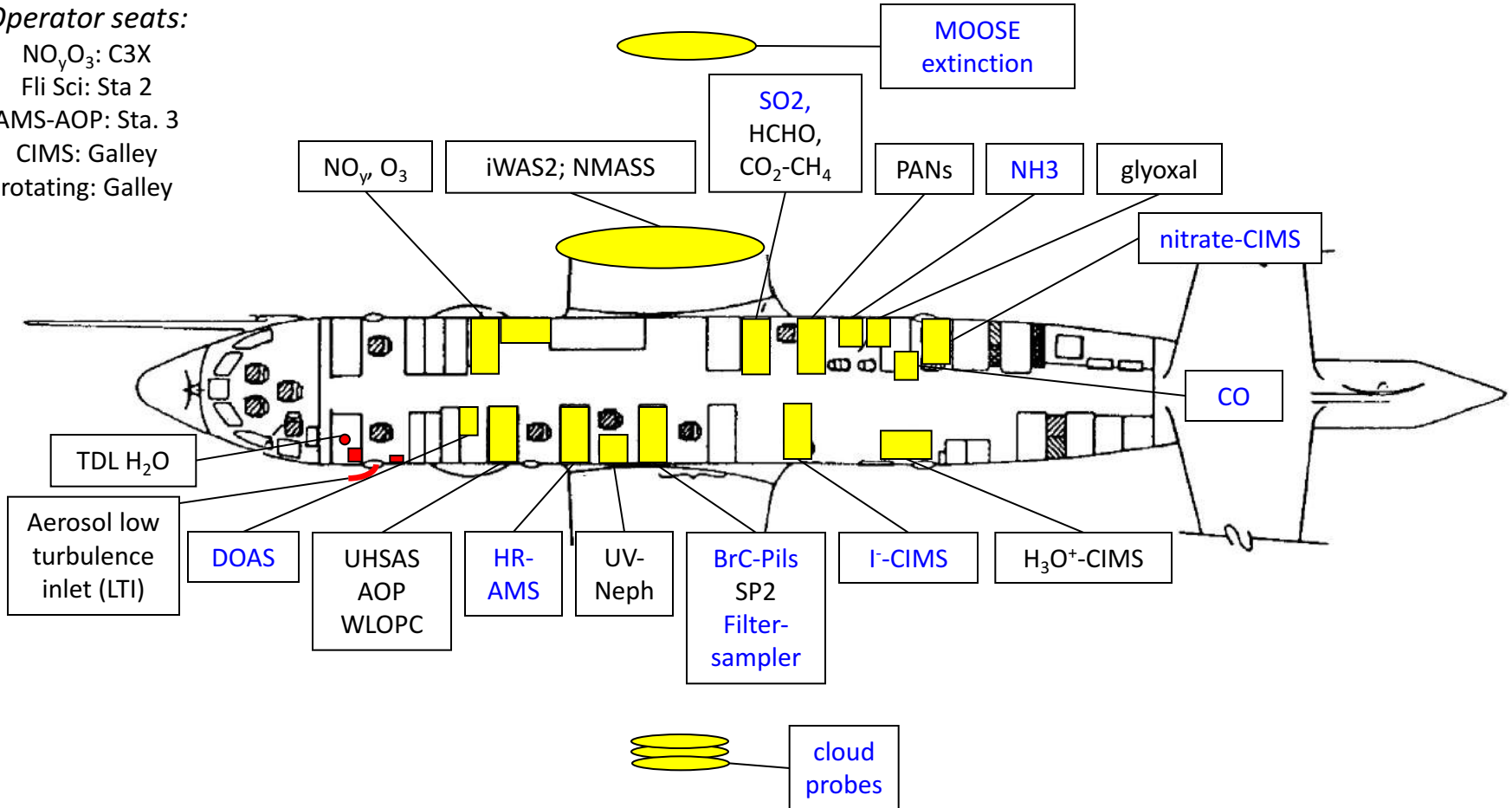


# N43RF Payload- FIREX 2019

NOAA-CSD preliminary version 08 06-20-2017

## Operator seats:

$\text{NO}_y\text{O}_3$ : C3X  
Fli Sci: Sta 2  
AMS-AOP: Sta. 3  
CIMS: Galley  
rotating: Galley



# Tentative Instrument List NOAA P3 2019

Ion	Abbreviation	Full name	Description	PI	Affiliation
	TDL H <sub>2</sub> O	Tunable Diode Laser water vapor	water vapor using open-path fast-response tunable diode laser absorption spectrometer	Dana Naeher	NOAA/AOC
	LTI	Low Turbulence Inlet	decelerating inlet to provide sample air to aerosol instruments in fuselage	Chuck Brock	NOAA/ESRL
&	WLOPC	White-Light Optical Particle Counter	supermicron aerosol number and sizes; samples from LTI	Chuck Brock	NOAA/ESRL
d	DOAS	Column NO <sub>2</sub> , HONO, HCHO, O <sub>3</sub> , SO <sub>2</sub>	trace gas columns 20km ahead of the aircraft	Jochen Stutz	UCLA
	CRD-AES	Cavity RingDown-Aerosol Extinction Spectrometer	total dry aerosol light extinction and extinction as f(RH); samples from LTI	Nick Wagner	NOAA/ESRL
	PSAP	Particle Soot Absorption Photometer	total aerosol light absorption by filter darkening; samples from LTI	Nick Wagner	NOAA/ESRL
	PAS	Photoacoustic Absorption Spectrometer	total aerosol light absorption by photoacoustics; samples from LTI	Nick Wagner	NOAA/ESRL
	UHSAS	Ultrahigh Sensitivity Aerosol size Spectrometer	counts and sizes 0.07-1.0 μm aerosol particles; samples from LTI	Chuck Brock	NOAA/ESRL
	HR-AMS	High resolution Aerosol Mass Spectrometer	Size resolved chemical composition of aerosol particles	Ann Middlebrook Katherine Hayden	Environ. Canada NOAA/ESRL
X	NO/NO <sub>2</sub> /NO <sub>y</sub> /O <sub>3</sub>	Nitrogen oxides and ozone	chemiluminescence detection with photolytic or catalytic conversion	Chelsea Thompson, Tom Ryerson	NOAA/ESRL
	SP2	Single-Particle Soot Photometer	soot particles number, size, and coating	Joshua Schwarz	NOAA/ESRL
	BrC-Pils	BrownCarbon-Particle into liquid sampler	absorption and concentration of water-soluble organic carbon	Rebecca Washenfelder	NOAA/ESRL
	Filter samples	Filter samples for I/SVOCs	GCxGC/TOF-MS (EI) and LC/MC	Kelley Barsanti Lindsay Hatch	UC Riverside
	j-values	Filter Radiometers	j <sub>NO2</sub> and j <sub>ozone</sub> using filter radiometers	Chelsea Thompson	NOAA/ESRL
	ISAF HCHO	In-situ airborne formaldehyde	formaldehyde (HCHO) using laser-induced fluorescence	Glenn Wolfe, Thomas Hanisco	NASA GSFC
	CH <sub>4</sub> , CO <sub>2</sub>	Picarro CH <sub>4</sub> , CO <sub>2</sub>	CO <sub>2</sub> and methane with IR laser absorption in a high-finesse cavity	Jeff Peischl, Tom Ryerson	NOAA/ESRL
	SO <sub>2</sub>	Sulfur dioxide	SO <sub>2</sub> using laser-induced fluorescence	Andrew Rollins	NOAA/ESRL
	PAN CIMS	PeroxyAcyl Nitrate CIMS	PANs using chemical ionization mass spectrometry with I <sup>-</sup> as reagent ion	Patrick Veres	NOAA/ESRL
	NH <sub>3</sub>	QC-TILDAS ammonia	ammonia using Quantum Cascade Tunable Infrared Laser Differential Absorption Spectroscopy	Jennifer Murphy	University of Toronto
	ACES	Airborne Cavity Enhanced Spectrometer	Glyoxal using Cavity Enhanced Absorption Spectroscopy	Kyle Zarzana, Steve Brown	NOAA/ESRL
	CO, H <sub>2</sub> O	Los Gatos CO	cavity enhanced absorption technique in high-finesse	Jeff Peischl	NOAA/ESRL

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Detailed instrument list/information available at: <http://esrl.noaa.gov/csd/projects/firex/>

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	CO, H <sub>2</sub> O	Los Gatos CO	cavity enhanced absorption technique in high-finesse	Jeff Peischl	NOAA/ESRL



Phase 1 of FIREX  
Missoula USFS Fire Science Lab  
Oct 1- Nov 15 2016

Detailed smoke emission characterization in a  
controlled environment

First Results

Over 60 people from 20 institutions in Missoula!  
9 NOAA funded projects



AMS  
ACES  
FTIR, PAX

I-CIMS  
NO3-CIMS  
Samplers



**Samplers:**  
Gas: GCxGC-ToF-MS (EI)  
Particle: GCxGC-ToF-MS (VUV)  
DI-MS, PILS



**Wind Tunnel:**

**Mini Chamber:** HR-AMS  
SP-HR-AMS  
CAPS, PASS  
CRD-PAS, SP2  
CO, CO2, O3

**Mixing Drum for BC and BrC:** SP2, CLAP, POPS  
PAX, SMPS  
WSOC-PILS



Burn Room

**Room Burns:**

BrC-PiLS  
BBCEAS  
CRDPAS  
NEPH



**Control Room:**

H3O+CIMS  
I-CIMS  
PAM  
GC-MS



**Viewing Room:**

PILS-ESI  
Ny, NO, sampler



CSU Chamber



I-CIMS



**Aerodyne Lab:**

LToF-AMS  
PTR-MS  
ECHAMP  
PAM  
CO, CO2, NOx, HCHO,  
CH4, C2H6, C3H8, ..

- Fuels:
- Ponderosa Pine and Lodgepole Pine
  - Subalpine Fir and Douglas Fir
  - Engelmann Spruce
  - Chaparral: Manzanita and Chamise
  - FASMEE: Subalpine Fir, Ft Stuart
  - additional fuels from SE US, peat, dung, sage, excelsior

Emission factors  
(EF) for over  
100 fires

- Conditions:
- realistic with duff, litter, logs and canopy of NW US fuels
  - separated burns for duff, litter, logs and canopy
  - N-content and BC/BrC ratio variation

Stack burns for emissions

Room burns for detailed smoke characterization

Smoke aging in simulation chambers

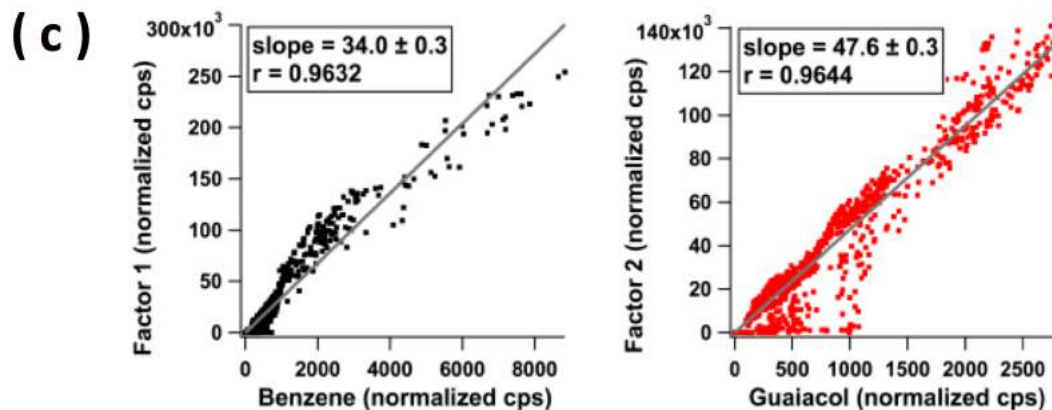
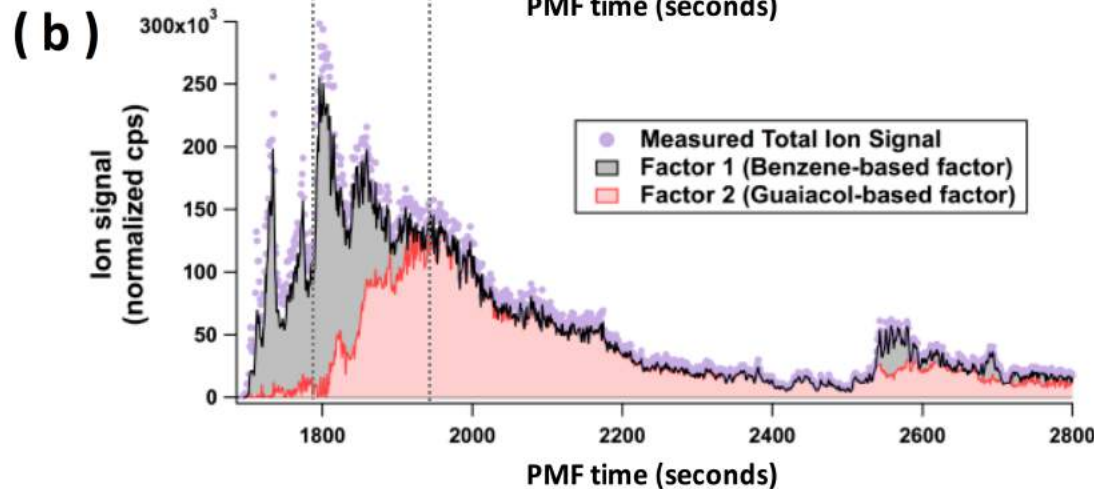
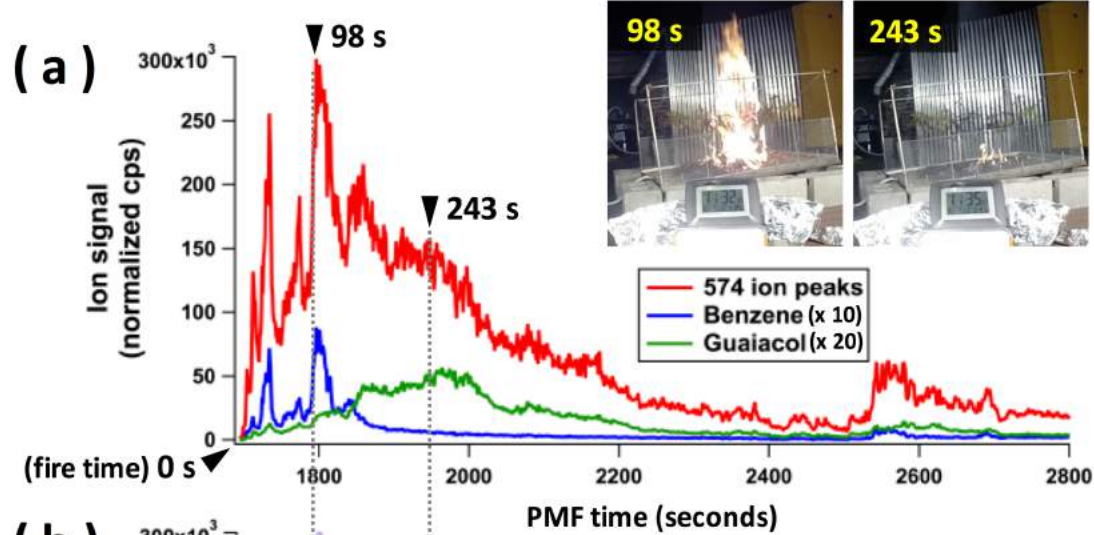


Most detailed smoke,  
emissions and chemistry  
characterization to date

Example: Volatile Organic Compounds (VOCs) with H3O+CIMS

- Different fire stages
- Different fuels

# Flaming versus Smoldering from Ponderosa Pine



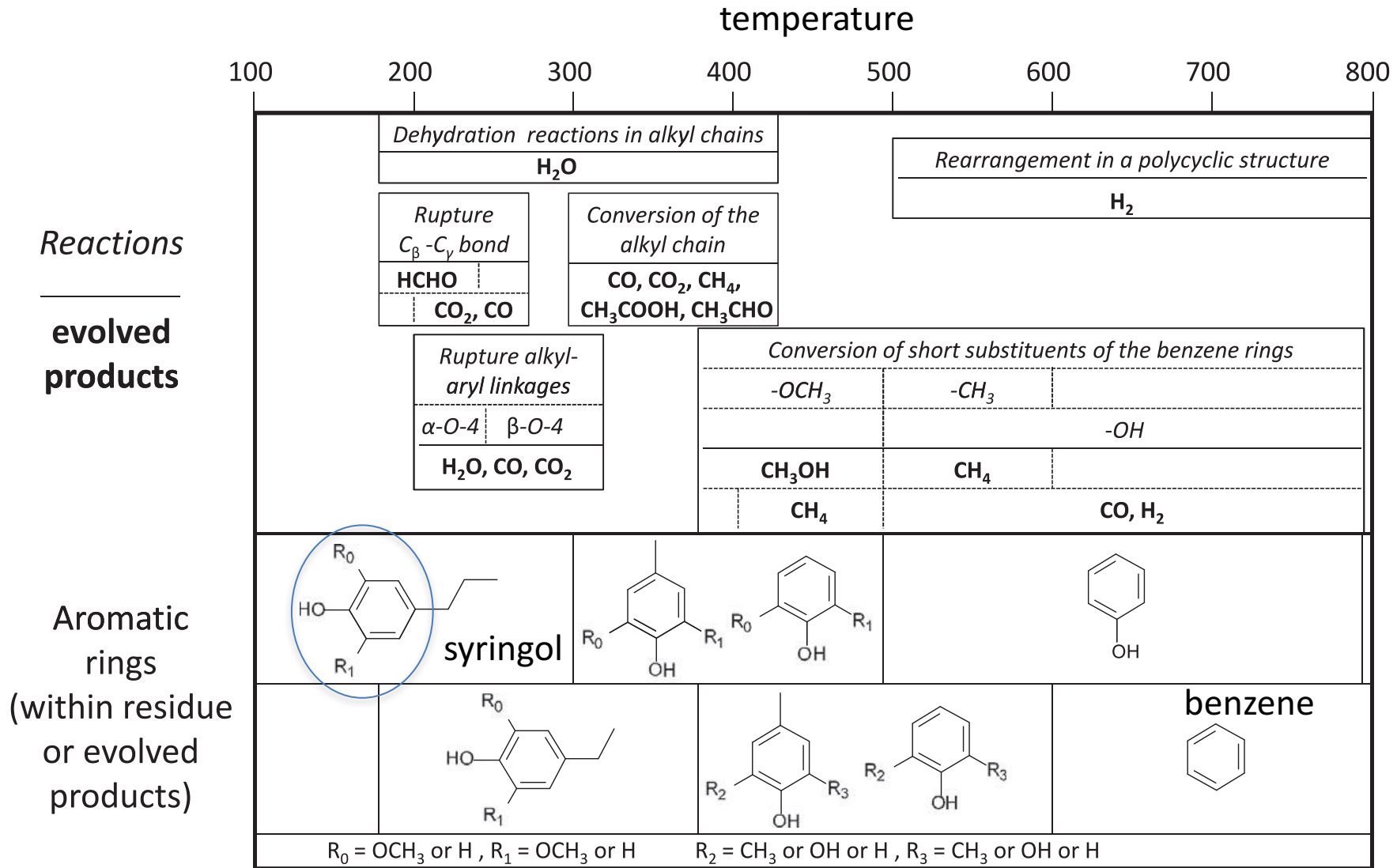
Example: Volatile Organic Compounds (VOCs) with H<sub>3</sub>O+CIMS

- Instrument detects VOCs by proton-transfer with H<sub>3</sub>O<sup>+</sup>
- >500 compounds detected

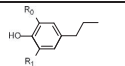
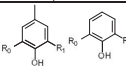


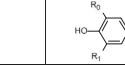
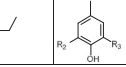
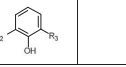

Positive Matrix Factorization statistical method determining contributions of compounds to the fire stages

Different compounds are emitted at different temps.

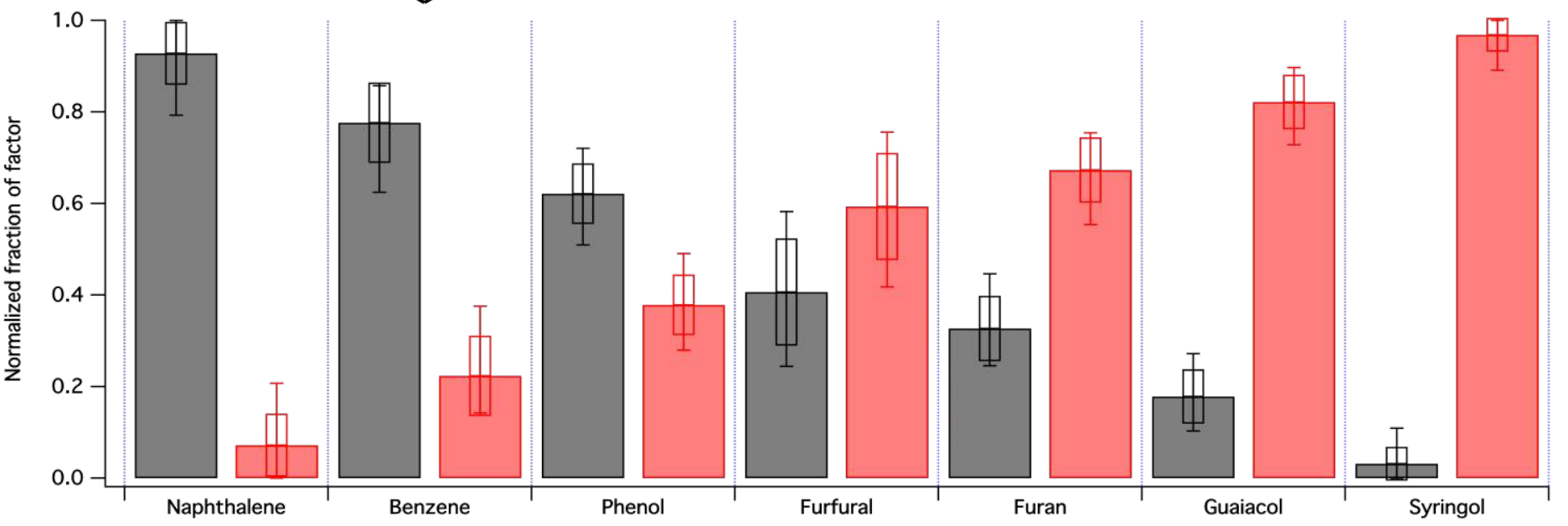
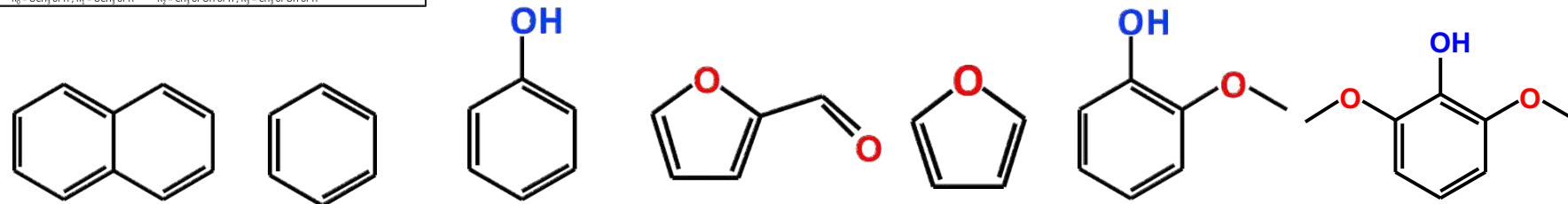
# Products from Lignin Pyrolysis



**B**

	100	200	300	400	500	600	700	800	
Reactions	Dehydration reactions in alkyl chains			Rearrangement in a polycyclic structure					
	H <sub>2</sub> O			H <sub>2</sub>					
	Rupture C-C bond		Conversion of the alkyl chain						
	HCHO		CO, CO <sub>2</sub> , CH <sub>4</sub> , CH <sub>3</sub> COOH, CH <sub>3</sub> CHO						
evolved products	Rupture alkyl-aryl linkages		Conversion of short substituents of the benzene rings						
	α-O-4		-OCH <sub>3</sub>		-CH <sub>3</sub>		-OH		
	β-O-4		CH <sub>3</sub> OH		CH <sub>4</sub>		CO, H <sub>2</sub>		
	H <sub>2</sub> O, CO, CO <sub>2</sub>		CH <sub>4</sub>						
Aromatic rings (within residue or evolved products)									
									
R <sub>0</sub> = OCH <sub>3</sub> or H, R <sub>1</sub> = OCH <sub>3</sub> or H		R <sub>2</sub> = CH <sub>3</sub> or OH or H, R <sub>3</sub> = CH <sub>3</sub> or OH or H		R <sub>2</sub> = CH <sub>3</sub> or OH or H, R <sub>3</sub> = CH <sub>3</sub> or OH or H					

# Products from Lignin Pyrolysis

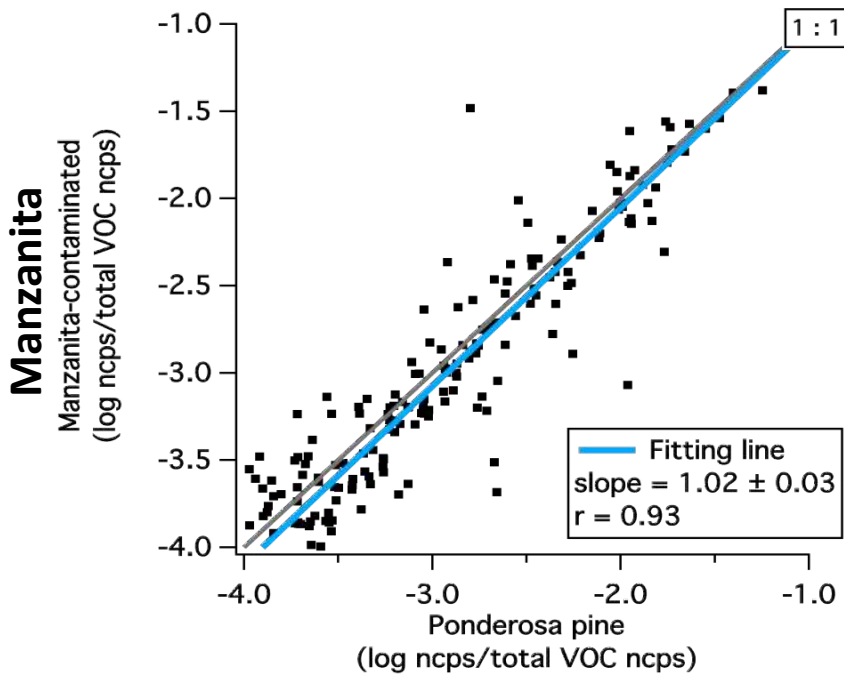


High temp.  
(~ 850 °C)

Biomass pyrolysis products

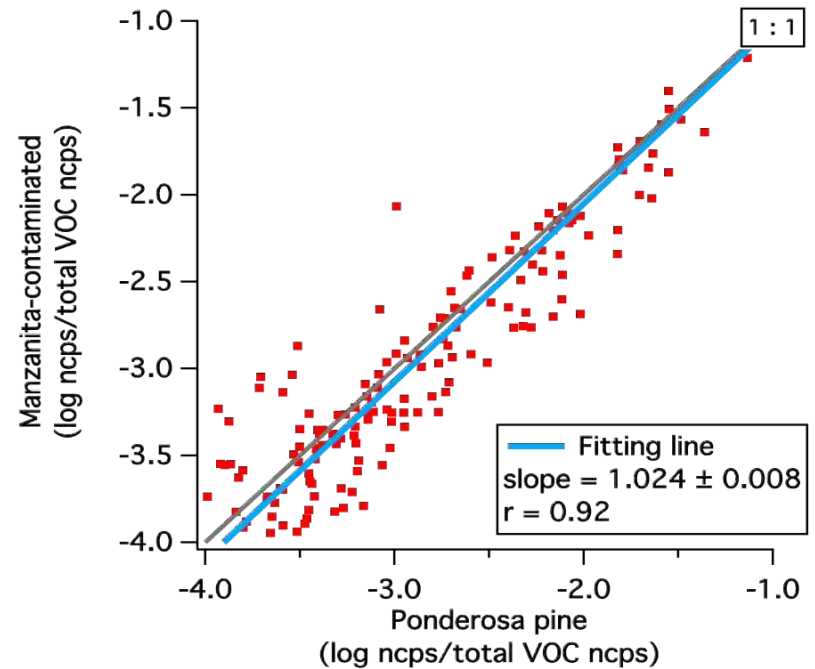
Low temp.  
(~ 200 °C)

# Ponderosa Pine versus Manzanita



Ponderosa pine

High-temperature



Ponderosa pine

Low-temperature

Different fuels emit similar compounds at similar temperatures



Over 60 people from 20 institutions in Missoula!

Photo: Abril Galang

Special thanks to NOAA AC4 for funding!





More FIREX details and list of partners

available at: <http://esrl.noaa.gov/csd/projects/firex/>

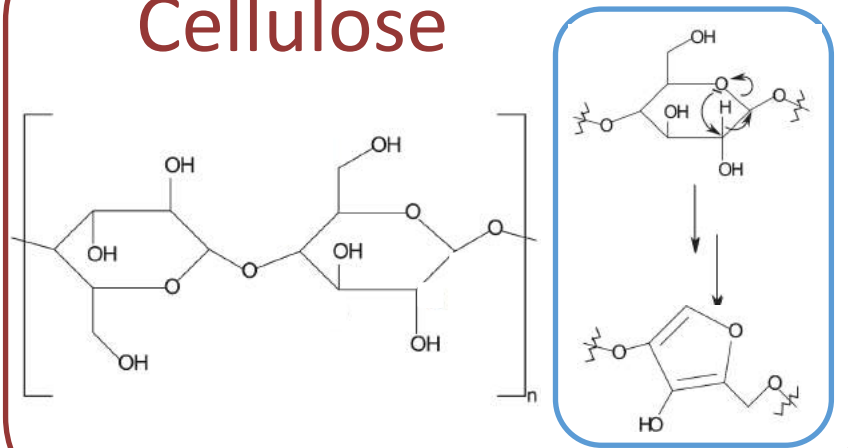
Please contact me for any FIREX related questions

and potential collaborations!!!

**Special thanks to NOAA AC4 for funding!**

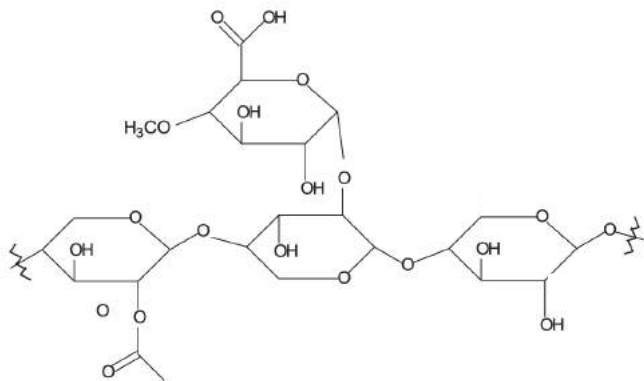
Many important VOCs we observe in fires reflect the polymer components of biomass

## Cellulose

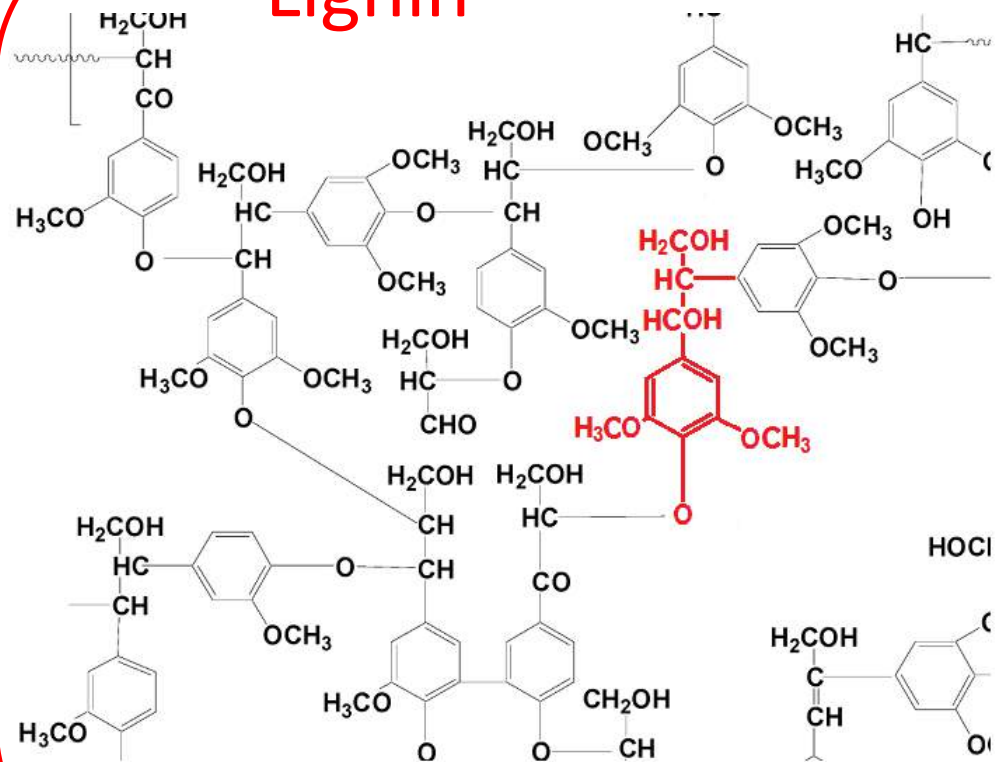


*Collard & Blin, Renew. & Sust. Energy Rev. (2014)*

## Hemicellulose



## Lignin



*Mohan et al., Energy & Fuels (2006)*