

4.005 OH-Radical Oxidation of Organic Species at the Air-Water Interface.

Presenting Author:

SHINICHI ENAMI, National Institute for Environmental Studies, Onogawa, Tsukuba, Ibaraki 305-8506, Japan, enami.shinichi@nies.go.jp

Abstract:

Products and intermediates of the prompt oxidation of a variety of carboxylic acids, including cis-pinonic acid and benzoic acid, initiated by gas-phase hydroxyl radicals, $\cdot\text{OH}(\text{g})$, at the air-water interface were detected by mass spectrometry in a novel setup under various experimental conditions. Exposure of sub-millimolar reactant aqueous microjets to ~ 10 ns $\cdot\text{OH}(\text{g})$ pulses from the 266 nm laser flash photolysis of $\text{O}_3(\text{g})/\text{O}_2(\text{g})/\text{H}_2\text{O}(\text{g})/\text{N}_2(\text{g})$ mixtures yielded an array of interfacial species, including peroxy radicals $\text{ROO}\cdot$ and hydroperoxides ROOH , that were unambiguously and simultaneously identified *in situ* by mass spectrometry. Our results are consistent with a Langmuir–Hinshelwood mechanism rather than an Eley-Rideal mechanism. We propose an interface-specific oxidation mechanism for the heterogeneous reaction of amphiphilic organic species (aq) with $\cdot\text{OH}(\text{g})$. The connection between the present fundamental findings and atmospheric chemistry is discussed.